

CO₂ gasification of microalgae (*N. Oculata*) – A thermodynamic study

Muflih Arisa Adnan^{1,2*} and Mohammad Mozahar Hossain²

¹Department of Chemical Engineering, Islamic University of Indonesia, Yogyakarta 55584, Indonesia

²Department of Chemical Engineering, King Fahd University of Petroleum & Minerals, Dhahran 31261, Saudi Arabia

Abstract. A new model of CO₂ gasification has been developed in the Aspen Plus. The potential of microalgae (*N. oculata*) for CO₂ gasification also has been investigated. The present gasification process utilizes the CO₂ at atmospheric pressure as the gasifying agent. The steam is also injected to the gasification to enhance the H₂ production. The composition of the producer gas and gasification system efficiency (GSE) are used for performance evaluation. It is found that the CO₂ gasification of microalgae produces a producer gas with a high concentration of CO and H₂. The GSE indicates that the process works at high performance.

1 Introduction

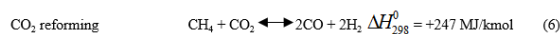
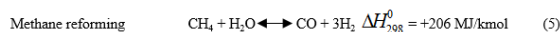
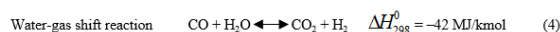
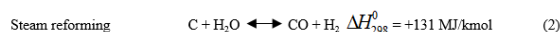
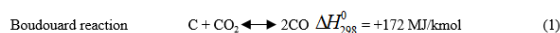
The increase of the fossil fuel consumption during the last decades causes a global warming. The renewable fuel has been considered as a potential solution of the corresponding problem. Among the available renewable resources, biomass is considered as a potential alternative resources with minimum negative impact due to its ability to reduce the CO₂ from the atmosphere and its low sulfur content [1].

Gasification is a potential biomass conversion process since it produces a producer gas which mainly consists of H₂, CO, CO₂ and CH₄ [2]. The gasifying agent has a considerable effect on the producer gas composition. [3], [4] and [5, 6] reported the injection of steam to the gasifier enhanced the H₂ production in the producer gas. The use of CO₂ as a gasifying agent attracts the attention of some researchers. For instance, [7] investigated the gasification of pine sawdust using CO₂ over Ni/Al precipitation catalyst. [8] studied the gasification of beech sawdust in the entrained flow reactor. [9] developed a model to investigate the performance of biomass gasification with CO₂ using fluidized bed gasifier. Recently, some research have focused on microalgae cultivation and processing because it offers positive environmental impacts [10-12]. To the best of our knowledge, only one literature discussing the CO₂ gasification of algae. [13] carried out CO₂ gasification of microalgae char using high-pressure thermogravimetric analyzer (HP-TGA).

2 Methodology

The minimization of Gibbs free energy is used for thermodynamic modeling of gasification process which involves solid, liquid and gas phases [14]. The process flow diagram of the gasification system can be

seen in Fig. 1. The biomass and the gasifying agents (CO₂ and H₂O) are fed to the gasifier separately. It is worth noting that the gasifier is modelled using two blocks: the DECOM block (RYield) and the Reduction block (RGibbs). A Fortran code is embedded in the DECOM block to convert the microalgae feed stock (non-conventional component) into conventional components. While the gasification reactions occur based on the method of Minimization of Gibbs Free Energy in the Reduction block. It is worth mentioning that during gasification process the following consecutive reactions occur in the gasifier:



The gasification products flow to the cyclone to separate the gaseous products (H₂, CO, H₂O, and CO₂) and solid products (ash and unconverted char). The gaseous products are fed to the reformer. The reformer products is sent to the CO₂ absorber to capture the CO₂ in the reformer products. The pure CO₂ from the absorber is then cooled and compressed to 523 K and 80 bar, respectively. The producer gas is expanded and cooled to 3 bar and 298 K, respectively. The ultimate and proximate analysis of the biomass, including the heating value of biomass are summarized in Table 1. The operating conditions of the gasification process are summarized in Table 2.

* Corresponding author: muflih.adnan@uii.ac.id

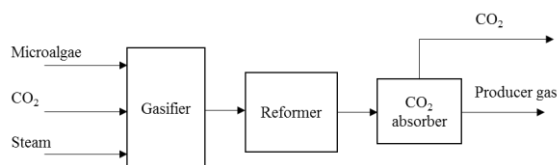


Fig. 1. Block diagram of the gasification process.

The performance of the gasification is evaluated in term of the dry composition of producer gas and gasification system efficiency (GSE) (Eq. (7)). One should note that the energy required for CO₂ absorption using amine is 3 MJ/kg CO₂ absorbed [15].

$$GSE = \frac{m_{pg} \times LHV_{pg} + E_{gen}}{m_{ma} \times LHV_{ma} + E_{req}} \quad (7)$$

Where m and LHV are mass flowrate and low heating value, respectively, while subscript pg and ma refer to the producer gas and the microalgae, respectively. The E_{gen} represents the total energy generated from the process while the E_{req} represents the total energy consumed by the process.

Table 1. Proximate and ultimate analysis of microalgae.

Microalgae (<i>N. oculata</i>) [11]	
Proximate (wt.%)	
Moisture	6.71
Volatile matters	78.94
Fixed carbon	7.95
Ash	6.4
Ultimate analysis (wt.%)	
C	47.50
H	6.15
O	46.35
N	n.r
S	n.r
Cl	n.r
High heating value (MJ/kg)	15.07

n.r : not reported

Table 2. Operating conditions in the simulation.

Inlet temperature of biomass, boiler feed water, and CO ₂	25°C
Temperature of steam entering gasifier	327°C
Temperature of gasifier	70°C
Temperature of reformer	800°C
Pressure in the all blocks	2000 kPa
Stream class	MIXCINC
Thermodynamic package	Peng-Robinson

It is worth noticing that the model validation of the gasification block is conducted using the same feedstock and operating conditions with the ones employed by [16]. The composition of the producer gas from the gasifier is used as the baseline for model

validation. It is clearly shown in Table 3 that the results of our simulation agree closely with the results reported by other author [16]. Indeed the relative error for H₂ and CO as the major producer gas component was less than 10%.

Table 3. Model validation. (Biomass CH_{1.4}O_{0.6}, CO₂/C = 0.5, T = 800 °C and P = 1 atm).

	Our simulation	Chaiwatanodom [16]	%Error
H ₂	30.45%	30.98%	1.75%
CO	63.13%	59.78%	5.31%
CO ₂	6.04%	9.01%	n.r
CH ₄	0.38%	4.30%	n.r

3 Results and discussion

A microalgae feedstock with mass flow rate equal to 100 kg/h is fed to the gasifier. In addition, a CO₂ and a steam with mass flow rate of 151 kg/h, and 62 kg/h, respectively, are selected as the gasifying agents. It is worth noticing that the gasification system consists of the gasifier, the reformer and the CO₂ absorber, as shown in Fig. 1.

It is clearly shown in Fig. 2a that the microalgae disappears while the amount of H₂, CO, CO₂, and CH₄ significantly increase in the product stream of the gasifier. This indicates that the main task of the gasifier is the conversion of the solid feed stock (e.g., microalgae) into the gaseous products (H₂, CO, CO₂, CH₄ and H₂O) with the help of the gasifying agents (H₂O and CO₂). The decrease of the amount of microalgae and H₂O indicates that the microalgae is decomposed into gaseous products through steam reforming reaction (Eq. (2)). This is also confirmed by the increase of the CO and H₂ in the gasifier product. One should notice that the methane formation reaction (Eq. (3)) also takes place during CO₂ gasification of microalgae since the CH₄ presents in the gasifier product. When one look at Fig. 2a, the amount of CO₂ in the gasifier products is higher than that of in the gasifier feed. This indicates that water-gas shift reaction (Eq. (4)) occurs in the gasifier. Indeed, this also can be attributed to the decrease of the H₂O amount in the gasifier product.

In the reformer unit, the gasifier products is further reacted to produce higher concentration of combustible gases (CO and H₂). Fig. 2b indicates that the amount of CO₂ and CH₄ in the reformer feed stream are higher than their counterparts in the reformer product stream. This can be attributed to the methane reforming reaction (Eq. (5)) and CO₂ reforming reaction (Eq. (6)). This is confirmed by the increase of the amount of CO and H₂ in the stream of the reformer product.

The amounts of the components in the feed of CO₂ absorber and in the producer gas stream are depicted in Fig. 2c. The amount of the CO₂ in the producer gas stream is lower than its counterpart in the stream of

CO₂ absorber feed. This indicates that the CO₂ absorber significantly reduces the amount of non-combustible gas in the producer gas stream. It is worth mentioning that the producer gas consists of 45% H₂, 45% CO, 6% CO₂ and 4% CH₄ (dry basis), while the GSE of the process is 68%.

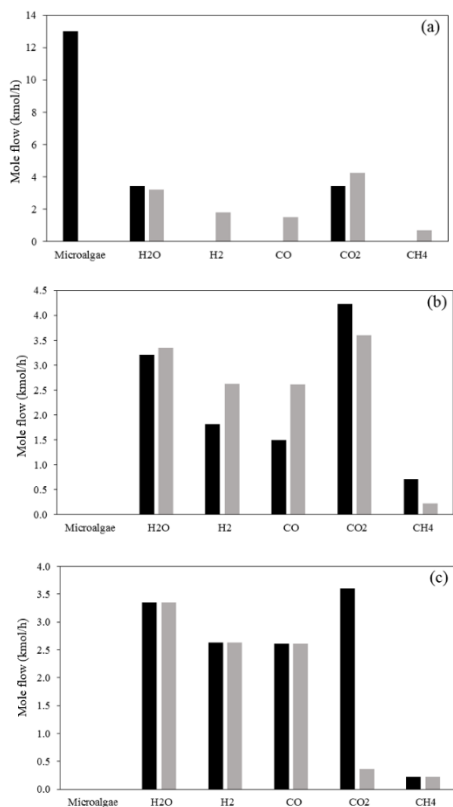


Fig. 2. Mole flow of the feed and product of (a) the gasifier, (b) the reformer, and (c) the CO₂ absorber. (feed: black; product: grey)

4 Conclusion

The production of producer gas with a high concentration of CO and H₂ via the CO₂ gasification of microalgae is simulated using Aspen Plus. The results show that the CO₂ gasification of microalgae can provide the producer gas with a high concentration of H₂ and CO. It is also reported that the GSE of the process is relatively high. In addition, high purity of CO₂ is also produced as the side-product of the process. Although the present model is developed only for the purpose of the preliminary study of the CO₂ gasification of algae. The results indicate that the CO₂ gasification of algae can potentially be a promising alternative of green technology with further optimization.

The research team acknowledges the financial support provided by King Abdul Aziz City for Science and Technology (KACST) to this research under KACST-TIC for CCS project no 03. The team also thanks the facilities and support provided by KFUPM.

References

- de Lasa, H., et al., *Catalytic Steam Gasification of Biomass: Catalysts, Thermodynamics and Kinetics*. Chemical Reviews, 2011. **111**(9): p. 5404-5433.
- Balu, E., U. Lee, and J.N. Chung, *High temperature steam gasification of woody biomass – A combined experimental and mathematical modeling approach*. International Journal of Hydrogen Energy, 2015. **40**(41): p. 14104-14115.
- Naqvi, M., et al., *An experimental study on hydrogen enriched gas with reduced tar formation using pre-treated olivine in dual bed steam gasification of mixed biomass compost*. International Journal of Hydrogen Energy, 2016. **41**(25): p. 10608-10618.
- Guan, Y., et al., *Steam catalytic gasification of municipal solid waste for producing tar-free fuel gas*. International Journal of Hydrogen Energy, 2009. **34**(23): p. 9341-9346.
- Adnan, M.A., et al., *Enhancement of hydrogen production in a modified moving bed downdraft gasifier – A thermodynamic study by including tar*. International Journal of Hydrogen Energy, 2017. **42**(16): p. 10971-10985.
- Adnan, M.A., et al., *Feed compositions and gasification potential of several biomasses including a microalgae: A thermodynamic modeling approach*. International Journal of Hydrogen Energy, 2017. **42**(27): p. 17009-17019.
- Garcia, L., et al., *CO₂ as a gasifying agent for gas production from pine sawdust at low temperatures using a Ni/Al coprecipitated catalyst*. Fuel Processing Technology, 2001. **69**(2): p. 157-174.
- Billaud, J., et al., *Influence of H₂O, CO₂ and O₂ addition on biomass gasification in entrained flow reactor conditions: Experiments and modelling*. Fuel, 2016. **166**: p. 166-178.
- Cheng, Y., Z. Thow, and C.-H. Wang, *Biomass gasification with CO₂ in a fluidized bed*. Powder Technology, 2016. **296**: p. 87-101.
- López-González, D., et al., *Kinetic analysis and thermal characterization of the microalgae combustion process by thermal analysis coupled to mass spectrometry*. Applied Energy, 2014. **114**: p. 227-237.
- Ali, S.A.M., S.A. Razzak, and M.M. Hossain, *Apparent kinetics of high temperature oxidative decomposition of microalgal biomass*. Bioresource Technology, 2015. **175**: p. 569-577.
- Kebelmann, K., et al., *Intermediate pyrolysis and product identification by TGA and Py-GC/MS of green microalgae and their extracted protein and lipid components*. Biomass and Bioenergy, 2013. **49**: p. 38-48.
- Soreanu, G., et al., *CO₂ gasification process performance for energetic valorization of microalgae*. Energy, 2017. **119**: p. 37-43.

14. Shabbar, S. and I. Janajreh, *Thermodynamic equilibrium analysis of coal gasification using Gibbs energy minimization method*. Energy Conversion and Management, 2013. **65**: p. 755-763.
15. Peters, L., et al., *CO₂ removal from natural gas by employing amine absorption and membrane technology—A technical and economical analysis*. Chemical Engineering Journal, 2011. **172**(2–3): p. 952-960.
16. Chaiwatanodom, P., S. Vivanpatarakij, and S. Assabumrungrat, *Thermodynamic analysis of biomass gasification with CO₂ recycle for synthesis gas production*. Applied Energy, 2014. **114**: p. 10-17.