

Effect of membrane on carbonation and carbon dioxide uptake of *Chlorella* sp.

Emma Suali*, Rosalam Sarbatly, SM Anisuzzaman, Farhana Abd. Lahin, Mohd Asyraf Asidin, Tiffanyca Jusnukin

Faculty of Engineering, Universiti Malaysia Sabah, 88400 Kota Kinabalu, Malaysia

Abstract Recent studies showed that as low as 5% CO₂ increased microalgae growth. However, common bioreactor operation resulted in low carbonation due to poor CO₂ mass transfer and this inhibited CO₂ uptake of microalgae. Although bubbling increases mass transfer of CO₂-O₂ exchange, preserving high dissolved CO₂ remains the most challenging of microalgae cultivation in bioreactor. In order to increase high dissolved CO₂ and CO₂-O₂ exchange, this study employed two types of membrane; hollow-fibre membrane for carbonation and hydrophobic membrane for deoxygenation. It was found that membrane increased carbonation from 20 % to 75 % when operated at control CO₂ concentration. The hollow-fibre membrane capable of creating as small as 2 mm bubble which effective for high carbonation. At the same time, it increased CO₂ uptake up to 85% in bioreactor. The hydrophobic membrane removed 43% O₂ from the bioreactor. Both membranes increased mass transfer of CO₂-O₂ exchange in bioreactor which stimulated microalgae growth.

Keywords: Membrane; microalgae; carbonation; deoxygenation; *Chlorella* sp.

1 Introduction

Carbonation of microalgae in bioreactor has reported since 1990s [1,2]. However, study of hollow fibre membrane rarely discussed although it increases CO₂ uptake [3-5]. Microalgae has potential to produce sufficient biomass if cultivated under CO₂ with a mean to remove photosynthetic O₂ [6]. Microalgae growth associated with CO₂ uptake, which increased under carbonation.

In general, bubbling provides aeration to microalgae and this resulted in low CO₂ mass transfer because of bubble formation [7-10]. This causes most CO₂ released to the bioreactor headspace and lessen its potential for microalgae growth. Microalgae lack in vascular tissue and an internal means to capture CO₂ outside watery media [11,12]. Thus, carbonation by membrane ensures CO₂ entrapped in the watery media. In addition, membrane capable to scatter CO₂ in the media and this increases potential use of CO₂ [13-15].

Another challenge of microalgae cultivation in bioreactor is high O₂ concentration. Microalgae produced O₂ as a side product of photosynthesis, which occurring simultaneously with CO₂ uptake. Oxygen reduces efficiency of continuous photosynthesis and thus interfere the CO₂ uptake in bioreactor [16,17]. Common cultivation uses aeration to remove O₂. However, aeration removes CO₂ and decreases carbonation efficiency due to unstable bubbling created during aeration. An optimum CO₂ – O₂ exchange required at least two membranes each acts as

CO₂ carrier and O₂ removal. Therefore, this study discussed hydrophobic hollow fibre membranes as O₂ removal in bioreactor and its effect on microalgae growth.

Carbonation affects media pH and microalgae tolerance on pH level varies among species [18,19]. This study has employed *Chlorella* sp. which known has high tolerance on pH level. The tolerance of *Chlorella* sp. was emphasize further in this study.

2 Materials and Methods

2.1 *Chlorella* sp and cultural condition

This study employed *Chlorella* sp from Borneo (found at 6 °N and 116 °E). In this study, the *Chlorella* was cultivated under lessened carbon source in an adjusted standard Jaworski medium. Thus, the medium solution has no contents of NaHCO₃, Cyanocobalamin, Thiamine HCl and Biotin. About 0.0401 kg/m³ (40 ppm) of *Chlorella* presents in the medium and used as a basis analysis on the *Chlorella* growth. The CO₂ gas enters Membrane 1 at controlled flowrates and microalgal media enter Membrane 2 for O₂ removal.

2.2 Membrane bioreactor set up

* Corresponding author: emma.suali@gmail.com

So far, no membrane can add and remove two different gases simultaneously. Thus, this study employed a hydrophobic-membrane with a model number UFS220 for carbonation and labelled as Membrane 1. The other membrane, 2 X 6 Radial Flow type superphobic membrane supplied by Liqui-Cel for O₂ removal and labelled as Membrane 2. Table 1 shows feature of Membrane 1 and Membrane 2 whereas others feature was published in previous work [10].

Figure 1 shows membranes position in the experimental setup. Membrane 1 connected to bioreactor 1 (BR1) and Membrane 2 connected to both bioreactors for O₂ removal. Two fluorescence white cool lamps lit up the bioreactors at 296 $\mu\text{E}/\text{m}^2\text{s}$ and a Lux meter with a model number LX-101 used to measure the light intensity.

Table 1. Characteristic of membranes

	Membrane 1	Membrane 2
Number of fibres	2400	3550
Fibre pore size (μm)	0.01 to 0.1	0.03 to 0.06
Length (mm)	495	146
Total surface area (m^2)	0.80	0.50
Shell diametre (mm)	60	55.60
Capacity (m^3/s) $\times 10^{-5}$	2.7	0.16 to 1.6
Fibre thickness (mm)	0.65	N/A
Shell side volume (m^3) $\times 10^{-3}$	0.43	0.08
Volume occupies by hollow fibre (m^3) $\times 10^{-3}$	2.5	N/A
Housing material	Polyvinyl chloride	Polyethylene
Membrane material	Polysulfone	Polypropylene

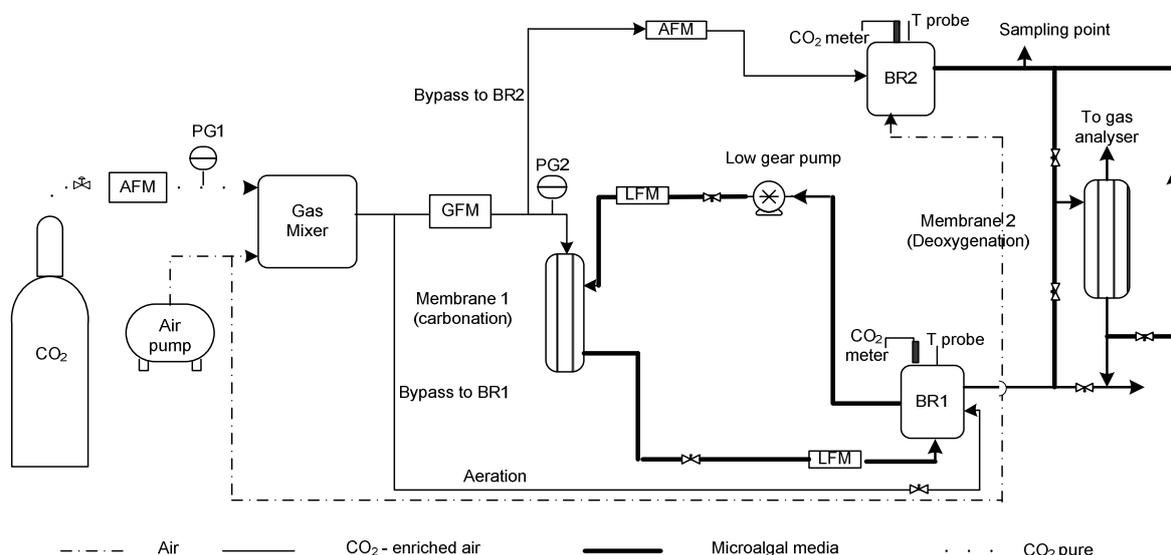


Fig.1. Membrane bioreactor set up

2.3 Carbonation and bubble measurement

The CO₂ feed was controlled from $0.49 \times 10^{-3} \text{ kg}/\text{m}^3$ to $3.11 \times 10^{-3} \text{ kg}/\text{m}^3$ for each experiment. The concentration of CO₂ in media and at the headspace of bioreactor was measured according to previous work [10]. The carbonic acid in media was estimated based on the total fraction of hydrogen ion to hydroxyl ions, which represented as pH reading [20].

The membrane created small bubbles in bioreactor, which was estimated according to Rodrigues and Rubio [21] method. The tubular bioreactor was labelled with numerical estimation from 0 cm to 100 cm as aid to the bubble estimation.

2.4 Oxygen removal by Membrane 2

Membrane 2 acted as O₂ removal prior to CO₂ depletion in the bioreactor. Membrane 2 is a superphobic

membrane, which have high resistance to water and microalgae attachment at the membrane walls.

2.5 CO₂ uptake and microalgae growth

The CO₂ uptake by microalgae is a measured of (a) dissolved CO₂, (b) carbonic acid in media, (c) CO₂ at the bioreactor headspace and (d) CO₂ in the membrane. The CO₂ meter measures CO₂ concentration from $1.49 \times 10^{-3} \text{ kg}/\text{m}^3$ to $1490 \times 10^{-3} \text{ kg}/\text{m}^3$ (1 ppm to 1490 ppm) in media and 0.1 % to 100 % in the bioreactor headspace. At the same time, a reading of microalgae growth, pH, and dissolved O₂ was recorded daily for three weeks. A UV-Vis spectrophotometer with UV-Vis bandwidth of 2.0 nm was used to measure growth of microalgae. The absorbance was recorded at a wavelength of 657 nm.

3 Results and Discussion

3.1 Effect of membrane integration

Figure 2a shows dissolved CO₂ in a bioreactor with and without membrane. At $0.49 \times 10^{-3} \text{ kg/m}^3$, the dissolved CO₂ was $0.1 \times 10^{-3} \text{ kg/m}^3$ in bioreactor without membrane. This equivalent to 20% of CO₂ feed. In comparison, the dissolved CO₂ in membrane integrated-bioreactor was $0.4 \times 10^{-3} \text{ kg/m}^3$. This equivalent to 82% of CO₂ feed and 4 times higher than in bioreactor without membrane. At $3.11 \times 10^{-3} \text{ kg/m}^3$, about 42 % and 29 % of CO₂ was dissolved in media for integrated and non-integrated bioreactor, respectively. Thus, further increase in CO₂ feed was considered as ineffective beyond $3.11 \times 10^{-3} \text{ kg/m}^3$.

The experimental result suggests that up to 82 % of CO₂ feed was delivered by membrane to the bioreactor and entrapped within the microalgae culture. Nearly 18 % was released into the headspace of the bioreactor, piled up within the membrane and converted to carbonic acid. From the perspective of Henry's law, the highest CO₂ could dissolve in water was approximated about 83 % of total supply (dissolved CO₂ = $0.8317 \times \text{CO}_2$ inlet) [22]. This means that 83 % of CO₂ was dissolved in water at atmospheric pressure and room temperature (28 °C). However, the equilibrium of CO₂ dissolves with CO₂ feed which is the solubility of CO₂ at natural conditions takes longer period to achieve compared with membrane-integrated bioreactor. This was a reason for drawback in microalgae cultivation with added CO₂. During carbonation, pH level was controlled at tolerable conditions for microalgae growth.

The acidic level of media in bioreactor was indicated by pH reading. From Figure 2b, pH level decreased as dissolved CO₂ increased. The natural conversion of CO₂ into carbonic acid in the presences of water caused the final reading of pH in the media as low as 4.82 [23]. It is desired to have high dissolved CO₂ compared to carbonic acid for microalgae cultivation. Microalgae ability that can use dissolved CO₂ without depending on carbon concentrating mechanism (CCM) increased CO₂ uptake in bioreactor. In comparison, carbonic acid conversion to useable carbon such as carbonate required CCM activity. Some microalgae have low CCM activity thus incapable to convert carbonic acid to useable carbon for growth. Microalgae can survive if less than 5% of carbonic acid in media [24,25]. Therefore, a small fraction of carbonic acid resulted from the reaction with water is sufficient for growth. Cultivation with more than 5% carbonic acid can cause acidic to the media.

Figure 2b shows that allowed CO₂ feed is less than $0.5 \times 10^{-3} \text{ kg/m}^3$ CO₂ to prevent extreme acidic condition. In overall, Figure 2 shows that carbonic acid increased with CO₂ feed. It was found that CO₂ conversion to carbonic acid agreed with Henry's law statement whereas the equilibrium constant of carbonic acid in water is approximated at 1.7×10^{-3} at room temperature [26]. Most of the CO₂ gas remains as CO₂ in water. The remaining CO₂ was scattered to another part of the membrane bioreactor. Thus, further study was conducted to evaluate the CO₂ distribution.

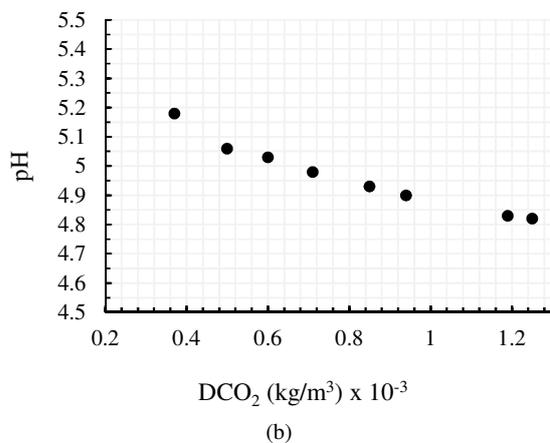
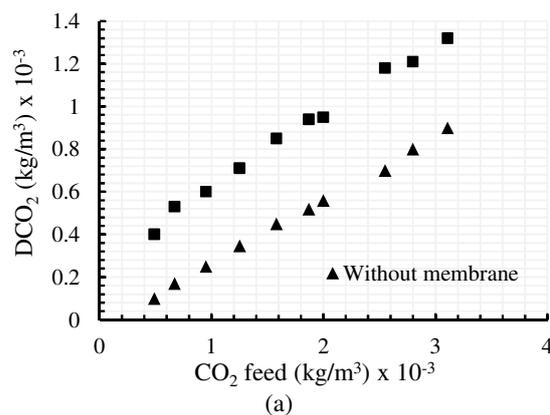


Fig.2. (a) Comparison of dissolved CO₂ in bioreactor; (b) pH

3.2 Distribution of CO₂ in bioreactor during carbonation

Carbon dioxide distributed in bioreactor mainly as dissolved gas in media, gas at the bioreactor headspace, gas piled up within the membrane and chemically converted to carbonic acid. Figure 3a, Figure 3b, Figure 3c and Figure 3d show the fraction of this CO₂, respectively. The membrane increased CO₂ delivery to media from 40 % to 80 % of CO₂ feed and thus reduced CO₂ at the bioreactor headspace up to 18 %. About 14 % amassed in the membrane and the smallest fraction, less than 1 % converted as carbonic acid in media.

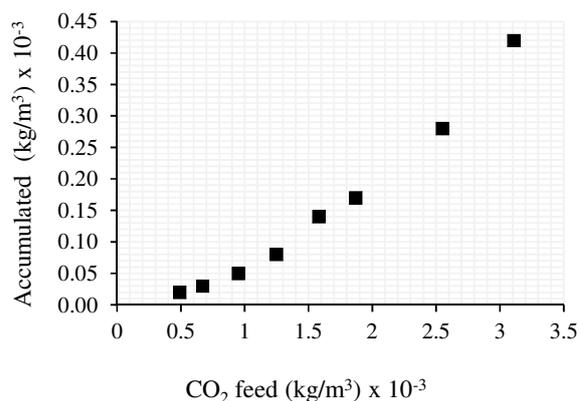
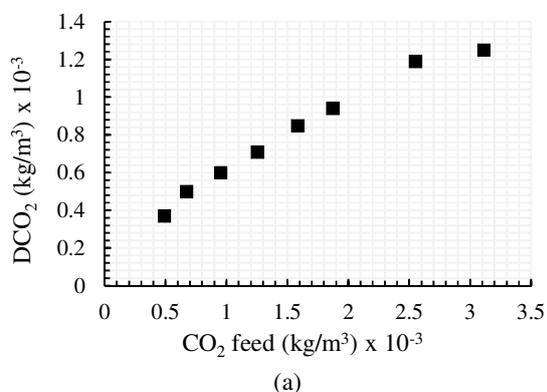
In Figure 3a, dissolved CO₂ increased as the CO₂ feed increased. However, low increment of dissolved CO₂ was found when fed with CO₂ above $2.5 \times 10^{-3} \text{ kg/m}^3$. This suggests that CO₂ should not feed beyond $2.5 \times 10^{-3} \text{ kg/m}^3$. There are three main reasons for the loss which were presented in Figure 3b, Figure 3c and Figure 3d.

In Figure 3b, the accumulated CO₂ within the membrane was $0.02 \times 10^{-3} \text{ kg/m}^3$ to $0.42 \times 10^{-3} \text{ kg/m}^3$ compared to the CO₂ feed which was 0.49×10^{-3} to $3.11 \times 10^{-3} \text{ kg/m}^3$. This equivalent to 4 % to 14 % of CO₂ feed. This shows that the higher the CO₂ feed, the higher possibility of the CO₂ accumulated within the membrane. This considered as a loss to the cultivation of microalgae. However, a simple flushing by air at the final stage of the carbonation lessens the gas build-up in the membrane.

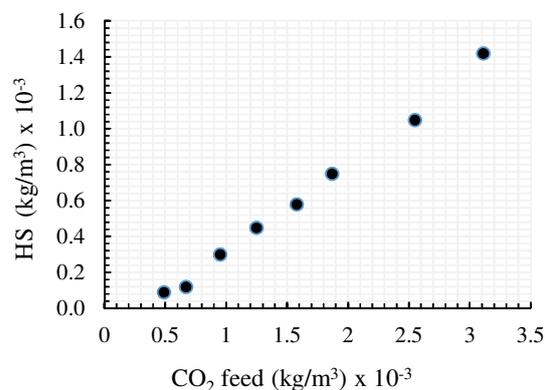
The remaining CO_2 was stored as gas at the bioreactor headspace. In Figure 3c, up to $1.42 \times 10^{-3} \text{ kg/m}^3$ of CO_2 feed was found in the bioreactor headspace. This equivalent to 0.46 % of CO_2 feed. The stability of bubble formation during carbonation has affected gas concentration at the bioreactor headspace although it lower than CO_2 piled-up in the membrane. The CO_2 at the headspace lessen the possibility of CO_2 uptake by *Chlorella*. It is possible to use CO_2 outside the watery media but takes longer period to achieve equilibrium between the gas and liquid phase. This affects microalgae growth. Thus, CO_2 dissolved into media have more benefit for microalgae growth compared to CO_2 outside the media.

The smallest fraction of the CO_2 feed was found as carbonic acid. The carbonic formation in microalgal-media has a slight difference compared with water. The converted CO_2 was in the range of $0.01 \times 10^{-3} \text{ kg/m}^3$ to $0.03 \times 10^{-3} \text{ kg/m}^3$ as shown in Figure 3d. This equivalent to 0.006 % to 0.02 % of the CO_2 feed. However, this amount caused acidic in microalgal media. Some microalgae species have high tolerance to the acidic environment. In this study *Chlorella* sp. showed no decrease in microalgae growth at low pH which suggested that *Chlorella* sp. might has high tolerance on low pH. Adding alkaline such as NaOH probably can reduce the effect of acidic condition in media. However, the effect of alkaline on pH was not evaluated in this study. In overall, the use of membrane to aid carbonation lower CO_2 amount at the headspace and within the membrane. It increased carbonation and ease CO_2 addition into the *Chlorella* sp. culture. The key to high increase in carbonation with the aid of membrane is the ability of membrane to control CO_2 feed and thus control over the bubble formation.

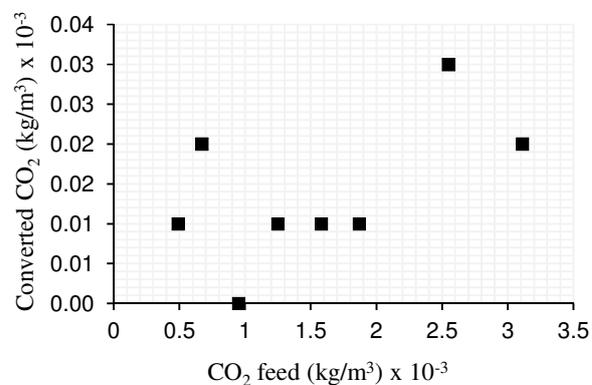
The bubbling is the apparent reason for low carbonation in bioreactor which aerated without membrane. Some study has reported that carbonation of microalgae culture better achieved by bubbling column than common aeration [3,27]. However, bubbling promotes low carbonation in bioreactor. The use of bubbling column increases microalgae growth up to 0.63 per day [28]. Thus, this study presents discussion on bubbling by membrane.



(b)



(c)



(d)

Fig.3. (a) Dissolved CO_2 ; (b) Amount of CO_2 in bioreactor headspace; (c) Accumulated CO_2 within the membrane; (d) Converted to carbonic acid

3.3 Relationship of carbonation and bubbling

Membrane distributes CO_2 in bioreactor and the key for achieving high dissolved CO_2 depends on membrane ability to control the size of the bubble. The bubble in bioreactor without membrane and aerated with standard air composition was found in the range of 5 mm to 8 mm in diameter. The bubble was in spherical shape. At constant flow rate, the bubble size was remained constant. The use of membrane resulted in small bubble ranges from 1 mm to 2 mm. Figure 4 shows bubble size compared with dissolved CO_2 . Figure 4 shows that the

bigger the bubble size, the lower amount of CO₂ dissolved. Flow rate also was found affected the bubble formation. Less than 2 mm bubbles observed at $0.3 \times 10^{-5} \text{ m}^3/\text{s}$. The dissolved CO₂ was found 70 % of the CO₂

feed. The bubble size increased with the gas feed. From $0.3 \times 10^{-5} \text{ m}^3/\text{s}$ to $0.7 \times 10^{-5} \text{ m}^3/\text{s}$ the bubble size was found from 1 mm to 2 mm and the dissolved CO₂ decreased from 70 % to 60 %.

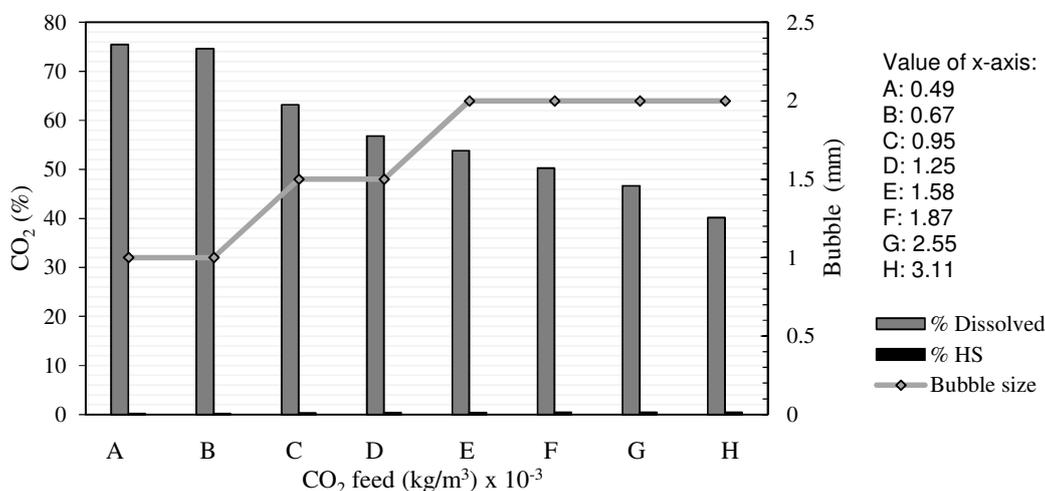


Fig.4. Effect of bubble on CO₂ concentration in media and bioreactor headspace

In continuous carbonation and cultivation, the bubble size decreased as the inlet media flow rate increased. At constant gas flow rate, which is $1.3 \times 10^{-5} \text{ m}^3/\text{s}$, the bubble size decreased from 2 mm to barely seen by eyes as a side effect of increasing liquid flowrate. The effect of liquid and gas flow rate in bioreactor was discussed in previous work [10].

As conclusion, the bubble size affected CO₂ in media and bubble created between 1 mm to 2 mm resulted in 70 % dissolved CO₂. When bubbled at 5 mm and above, the dissolved CO₂ was found less than 30 %. The large bubble created without membrane indicates that membrane plays a major effect on the bubbling formation. In addition, the inlet flow rate affected size of bubble. This result agreed with findings in bubbling column [29-31]. The instability of inlet flow rate and high Reynolds number creates large bubbles. The large bubble fetched up the CO₂ gas and escaped into the bioreactor headspace. Logically, microalgae only uses CO₂ that is in the water because of its cell phycology that does not have vascular tissue [32]. Thus, the CO₂ released into the headspace lessened the potential of CO₂ uptake by microalgae. The relation of carbonation and CO₂ uptake was measured further in this study.

3.4 Membrane carbonation and CO₂ uptake

The purpose of this study is to examine the range of tolerable carbonation or dissolved CO₂ by *Chlorella* sp. This work also aims to find a way to increase CO₂ uptake of *Chlorella* sp. Preliminary study shows that CO₂ uptake was in the range of 20 % to 85 % regardless of CO₂ feed. The highest CO₂ uptake was recorded at 2 mm bubble, followed by 1 mm. The lowest CO₂ uptake was when the cultivation was bubbled at 1 mm. It shows that at 1 mm,

Chlorella sp. used only 20 % of the total CO₂ feed. At the same time, the specific growth was recorded as 0.17 per day, which is the lowest among bubble size. However, this value higher compared to fast-growing grass which in the range of 0.09 d^{-1} to 0.59 d^{-1} [33].

The uneven distribution of CO₂ in media was identified as a main reason for low CO₂ uptake. The distribution of CO₂ bubbled at 1 mm was evaluated with the pH measure of the media. The pH was measured at the bottom, middle and surface of the media. Less than 0.2 % of CO₂ feed at the bioreactor headspace shows that most CO₂ entraps in the media. The *Chlorella* sp. growth which is 0.17 per day was due to insufficient CO₂ in the bioreactor. In overall, the CO₂ uptake by microalgae is best achieved when carbonized with bubble size approximately 2 mm where the CO₂ uptake was up to 85 %. This resulted in *Chlorella* sp. specific growth of 0.21 d^{-1} . This show that 2 mm was effective for both biomass and CO₂ uptake.

The CO₂ uptake in bioreactor associated with O₂ produced in the bioreactor. The total O₂ produced in bioreactor was 30 % and the dissolved O₂ was 18 %. The higher the growth rate, the higher the O₂ produced in the bioreactor. The O₂ generation also indicates the efficiency of CO₂ uptake. In ideal photosynthetic theory, each molecule of CO₂ produces one molecule of O₂. The O₂ in bioreactor headspace is in balance with the dissolved O₂. The presence of dissolved O₂ in the media causes photorespiration, thus preventing the photosynthetic enzyme to drive CO₂ uptake by microalgae. As low as 20 % of O₂ in media enough to hinder the photosynthesis cycle [14]. The O₂ reacts with RubisCO and produces phosphoglycolate, which inhibits the enzymes of photosynthesis.

The main factor of affecting CO₂ uptake was the bubbling within the bioreactor and the preliminary microalgae density in media. The density of microalgae affects light penetration inside the bioreactor. In extremely low microalgae density, light was easy to reach to each microalgae cell. However, the CO₂ uptake was found less compared to denser *Chlorella* sp. concentration. In addition, at extremely dense microalgae cell, more O₂ was produced. The produced O₂ inhibits the continuous photosynthesis cycle. Large bubbles in media capable of removing the dissolved O₂. Thus, the O₂ generated does not prevent the photosynthesis of *Chlorella* sp. as most of the dissolved O₂ released into the bioreactor headspace. The average ratio of dissolved O₂ to O₂ in bioreactor headspace was 4:1. Other means of removing O₂ without creating large bubble is applying hydrophobic membrane instead of aeration. Thus, preliminary effect of secondary membrane was discussed in this study.

3.5 Effect of secondary membrane to O₂ removal and *Chlorella* sp. growth

The Membrane 2 lessened O₂ in media from 8.5 x 10⁻³ kg/m³ to 4.9 x 10⁻³ kg/m³ at third week of cultivation. Another set of experiment which ran simultaneous also shows similar result. The BR1 which integrated with Membrane 1 was bubbled at 1 mm to 2 mm and BR2 without Membrane 1 ran at 5 mm to 10 mm. Both bioreactor were equipped with Membrane 2. About 41 x 10⁻³ kg/m³ of CO₂ was depleted in BR1. This equivalent to 89 % of the CO₂ feed. The overall dissolved O₂ and O₂ at the bioreactor headspace was about 24 % and 12 %, respectively. In comparison, BR2 shows 5 % and 21 % depletion of dissolved O₂ and O₂, respectively. The large bubble size in BR2 has aided the release of O₂ from the media. However, this resulted in low growth of *Chlorella* sp. which was only up to 0.19 per day (specific growth rate).

As a conclusion, the secondary membrane increased CO₂ uptake of *Chlorella* sp. The secondary membrane has no direct effect on the CO₂ uptake. However, both membranes have improved CO₂ uptake and O₂ removal from the bioreactor, thus, increased the *Chlorella* sp. growth.

4. Conclusion

This study presents an evaluation of membranes to increase carbonation of microalgae media. About 82 % of CO₂ feed was successfully delivered to the media with the aid of Membrane 1 and this resulted in low pH of media. The membrane also distributed CO₂ as dissolved CO₂ in media, piled up in membrane, accumulated at headspace and converted to carbonic acid. This study suggested that 2 mm bubble size is optimum for high carbonation. Finally, the secondary membrane has indirectly aided in increasing the CO₂ uptake by removing O₂ from the media.

Acknowledgements

This study was supported by FRGS grant no. FRG0416-TK-1/2015 and UMS grant with grant no. GUG0052-TK-2/2016 and GUG0050-TK-2/2016.

References

1. Y. K. Lee, H.K Hing, Appl. Microbiol. Biotechnol. **31**, 298-301 (1981)
2. S. Abinandan, S. Shanthakumar, 3 Biotech. **6**, 1-9 (2016)
3. R. Putt, M. Singh, S. Chinnasamy, K.C. Das, Bioresour. Technol. **102**, 3240-3245 (2011)
4. R. Sarbatly, E. Suali, F.A. Lahin, C.K. Chiam, In: *Advances in Bioprocess Technology*, Springer International Publishing, 371-386 (2015)
5. F.V. Winck, D.O.P. Melo, D.M. Riano-Pachon, M.C.M. Martins, C. Caldana, A. F. G. Barrios, Front. Plant Sci. **7**, 43 (2016)
6. S. P. Singh, P. Singh, Renew. Sust. Energ. Rev. **38**, 172-179 (2014)
7. H. S. Fogler, V.K.Verma, Chem. Eng. Sci. **26**, 1391-1400 (1971)
8. A. P. Carvalho, F.X. Malcata, Biotechnol. Prog. **17**, 265-272 (2001)
9. C. Vejrazka, M. Janssen, G. Benvenuti, M. Streefland, R.H. Wijffels, Appl. Microbiol. Biotechnol. **97**, 1523-1532 (2013)
10. R. Sarbatly, E. Suali, Algal. Res. **5**, 274-282 (2014)
11. L.Tomaselli, In: *Handbook of microalgal culture*, Biotechnol. Appl. Phycol., Blackwell Science Ltd, pp. 565 (2014)
12. J.A. Raven, J. Beardall, J. Exp. Bot. **67**, 1-13 (2016)
13. A. Contreras, F. Garcia, E. Molina, J.C. Merchuk, Biotechnol. Bioeng. **60**, 317-325 (1998)
14. F.C. Rubio, F. G. Fernandez, J.A. Perez, F.G. Camacho, E.M. Grima, Biotechnol. Bioeng. **62**, 71-86 (1999)
15. E. Molina, J. Fernandez, F.G. Acien, Y. Chisti, J. Biotechnol. **92**, 113-131 (2001)
16. C. Brindley, C., F.G. Acien, J.M. Fernandez-Sevilla, Biotechnol. Bioeng. **106**, 228-237 (2010)
17. M.Ota, Y. Kato, M. Watanabe, Y. Sato, R.L.Smith, S. Rosello-Sastre, C. Posten, H. Inomata, Bioresour. Technol. **102**, 3286-3292 (2011)
18. J.C. Goldman, Y. Azov, C.B. Riley, M.R. Dennett, J. Exp. Mar. Biol. Ecol. **57**, 1-13 (1982)
19. E. Touloupakis, B. Cicchi, A.M.S Benavides, G.Torzillo, Appl. Microbiol. Biotechnol. **100**, 1333-1341 (2016)
20. C.Y.Chen, E.G. Durbin, Mar. Ecol. Prog. Ser. **109**, 84-93 (1994)
21. R.T. Rodrigues, J. Rubio, Miner. Eng. **16**, 757-765 (2003)
22. M.L. Davis, S.J. Masten, *Principles of environmental engineering and science*, McGraw-Hill, New York (2004)
23. J. Thielmann, N.E. Tolbert, A. Goyal, H.Senger, Plant. Physiol. **92**, 622-629 (1990)

24. S.A. Schoenberg, R. Benner, A. Armstrong, P. Sobecky, R.E Hodson, *Appl. Environ. Microbiol.* **56**, 237-244 (1990)
25. R.C.D.Goswami, N. Kalita, M.C. Kalita, *Ann. Biol. Res.* **3**, 499-510 (2012)
26. J.M.Allen, W.X. Balcavage, B.R.Ramachandran, A.L. Shroud, *Environ. Toxicol. Chem.* **17**, 1216-1221 (1998)
27. A.S.Miron,C.F. Garcia, A.C. Gomez, E.M. Grima, Y. Chisti, *AIChE Journal*, **46**, 1872-1887 (2000)
28. G.Najafpour, N. Bakuei, G. Amini, M. Jahanshahi, *Indian J. Chem. Technol.* **22**, 20-25 (2016)
29. S.V.Gnyloskurenko, T. Nakamura, *Mater. Trans.* **44**, 2298-2302 (2003)
30. S.M. Walke, V.S. Sathe, *Int. J. Chem. Eng. Appl.* **3**, 25-30 (2012)
31. S.A. Sulaiman, N.Z.Z. Kamarudin, *J. Appl. Sci.* **12**, 2464-2468 (2012)
32. V.B.Rastogi, *Modern Biology*. Pitambar Publishing (1997)
33. E. Suali, R. Sarbatly, S.R.M. Shaleh, *International Conference on Applied Energy*, Jul 5-8, Suzhou China (2012)